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A new lignan glycoside from the rhizomes of *Imperata cylindrica*

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A new lignan glycoside, 6-acetyl-1-[1,3-(4,4'-dihydroxy-3,3'-dimethoxy- β -truxinyl)- β -D-fructofuranosyl]- α -D-glucopyranoside (**1**), named impecyloside, was isolated from the rhizomes of *Imperata cylindrica*. The structure of the compound was determined by spectroscopic data including FABMS, UV, IR, ¹H NMR and ¹³C NMR (DEPT) and 2D NMR (COSY, HSQC, HMBC).

Keywords: *Imperata cylindrica*; Impecyloside; Gramineae; Lignan glycoside

1. Introduction

Imperata cylindrica (Gramineae) is an aggressive, rhizomatous, perennial grass widely distributed in East Asia. The rhizomes of this plant have traditionally been used as a diuretic, an anti-inflammatory and an antipyretic agent in Korean herbal medicine.¹ Also, neuroprotective compounds from MeOH-extract have been reported.² Previous phytochemical studies on the rhizomes of *I. cylindrica* have resulted in the isolation of various compounds such as arundoin, cylindrin, cylindol, cylindrene, graminone and siderin.^{3–6} This paper describes the isolation and identification of a new β -truxinic acid derivative with an acetylated sucrose moiety from the rhizomes of *I. cylindrica*.

2. Results and discussion

A new β -truxinic acid derivative was isolated as a yellow amorphous powder from methanol, exhibiting an UV absorption maximum at 212 nm. The compound was freely soluble in water and in methanol. The molecular formula, C₃₄H₄₀O₁₈, was deduced from negative HRFAB-MS at *m/z* 735.2139 [M – H][–]. The IR absorption bands at 3400, 2937, 1730, 1600 cm^{–1} were due to hydroxyl, alkane, ester carbonyl and aromatic functions. In the ¹H NMR spectrum, two methoxy proton signals were observed at δ 3.61 and 3.73, along with six aromatic protons [δ 6.76 (br s, H-2), 7.03 (d, *J* = 8.0 Hz, H-5), 6.94 (br d, *J* = 8.0 Hz, H-6), 6.88 (br s, H-2'), 6.97 (d, *J* = 8.0 Hz, H-5'), 6.71 (br d, *J* = 8.0 Hz, H-6')] suggesting the presence of a dual 1,3,4-trisubstituted phenyl moiety. There were some oxygenated methine

and methylene signals of carbohydrate moieties in the region from δ 4.22 to 6.18 including one anomeric proton doublet at δ 6.18 with a coupling constant (*J*) of 3.2 Hz, an indicating α -glucosyl form.⁷ In the aliphatic region, methine proton signals at δ 4.80 (H-7'), 4.57 (H-7), 4.49 (H-8), and 4.37 (H-8') from the cyclobutane ring and acetyl-CH₃ proton signal at δ 1.88 were observed. The ¹³C NMR spectrum indicated thirty-four carbon signals. The multiplicity of each carbon was determined using DEPT experiment. In the low field region, three carbonyl signals at δ 171.1 (C-1'''), 171.9 (C-9') and 174.5 (C-9) were observed. In the aromatic region, six methine carbons [δ 112.7 (C-2, 2'), 115.9 (C-5'), 116.1 (C-5), 121.1 (C-6'), 121.5 (C-6)], two quaternary carbons [δ 130.3 (C-1'), 130.5 (C-1)] and four oxygenated quaternary carbon [δ 146.6 (C-4'), 146.7 (C-4), 148.1 (C-3'), 148.3 (C-3)] signals due to the dual 1,3,4-trisubstituted phenyl moiety structure were observed. Twelve oxygenated carbon signals of a carbohydrate moiety were observed from δ 63.3 to 109.6 including two anomeric carbon signals at δ 94.2 (C-1''') and 109.6 (C-2'') whose chemical shifts indicated the carbohydrate to be composed of a glucopyranose with some acetylated portion and a fructopyranose. In the aliphatic region, four methine carbon signals at δ 43.9 (C-8), 44.5 (C-8'), 44.8 (C-7') and 45.8 (C-7) of a cyclobutane ring, and an acetyl-CH₃ carbon signal at δ 20.7 (C-2''') were observed. In the HMBC spectrum, the anomeric proton signal of glucopyranosyl moiety at δ 6.18 showed correlations with an oxygenated carbon signal of the fructopyranosyl moiety at δ 109.6 (C-2''), indicating the oligosaccharide to be a sucrose [β -D-fructofuranosyl-(2 \rightarrow 1)- α -D-glucopyranose]. Two carbonyl carbon

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signals at δ 174.5 (C-9) and 171.9 (C-9') showed correlations with oxygenated methine proton signals of the fructopyranosyl moiety at δ 5.26 (H-1'') and 5.68 (H-3''), respectively. These results suggested ester bond formation at C-1'' and C-3'', confirmed by downfield shifts of the two proton signals owing to the esterification effect, as H-1'' and H-3'' signals of D-fructopyranose were usually observed at ca δ 3.60 and 4.10, respectively.^{8,9} Another acetyl carbonyl carbon signal at δ 171.1 (C-1''') showed a cross peak with an oxygenated methylene proton signal at δ 4.93 of the glucopyranosyl moiety (H-6''), meaning the hydroxy of C-6'' was acetylated, also confirmed by the downfield shift of the proton signal; typically observed at ca δ 3.70 to 3.80.^{8,9} The large coupling constants ($J_{7,7'} = 9.6$ and $J_{8,8'} = 11.2$ Hz) of methine protons of the cyclobutane ring resulted in *cis* configuration for pair H-7/H-7' and H-8/H-8'. Also, the coupling constants ($J_{7,8} = 5.2$ and $J_{7,8'} = 5.2$ Hz) of the methine protons led to inference of *trans* configuration for the H-7/H-8 and H-7'/H-8' pairs.⁹ Therefore, the structure of compound **1** (Figure 1) was determined to be 6-acetyl-1-[1,3-(4,4'-dihydroxy-3,3'-dimethoxy- β -truxinyl)- β -D-fructofuranosyl]- α -D-glucopyranoside, named impeccyloside.

The aglycone moiety of this glycoside, β -truxinic acid, and its sucrosyl glycoside have been isolated from tougucuo (*Incarvillea sinensis*) and oats, respectively.^{10,11} However, no β -truxinic acid derivatives have been reported from *Imperata cylindrica*. Also, organic solvents causing the acetylation reaction during extraction, fractionation and column chromatography procedure were not used, so impeccyloside must be a natural compound, rather than artificial. The very high anti-inflammatory activities of β -truxinic acid and its derivative compounds have been previously reported.¹¹

3. Experimental

3.1 General experimental procedures

Optical rotation was recorded on a JASCO P-1010 digital polarimeter (Tokyo, Japan). UV spectra were measured on a Shimadzu UV-1601 (Kyoto, Japan). The IR spectrum was obtained with a Perkin Elmer Spectrum One FT-IR spectrometer, CaF₂ window in MeOH (Buckinghamshire, UK). FAB-MS data were recorded on a JEOL JMS-700 (Tokyo, Japan). ¹H NMR (400 MHz), ¹³C NMR (100 MHz) and 2D NMR spectra were recorded on a Varian Unity Inova AS-400 FT-NMR spectrometer (California, USA). Pyridine-*d*₅ with TMS as internal standard was purchased from Sigma (St. Louis, MO, USA).

3.2 Plant material

The rhizomes of *Imperata cylindrica* were purchased at Kyungdongmart, a herbal drug store in Seoul, Korea and identified by Professor Dae-Keun Kim, College of Pharmacy, Woosuk University, Jeonju, Korea. A voucher specimen (KHU050731) is reserved at the Laboratory of Natural Products Chemistry, Kyung Hee University, Suwon, Korea.

3.3 Extraction and isolation

Dried rhizomes of *I. cylindrica* (1.5 kg) were extracted with 80% aqueous MeOH (2 L \times 3) at room temperature. The extracts were partitioned with water (1 L), EtOAc (1 L \times 3) and *n*-BuOH (0.8 L \times 3), successively. The concentrated *n*-BuOH extract (20 g) was applied to a Diaion-HP20 column (12 \times 60 cm) with a gradient eluting of H₂O/MeOH (1:0 \rightarrow 0:1) and the obtained fractions were checked by TLC. The 80% MeOH fraction [2 g, V_e/V_t (elution volume/total volume) 0.70–0.80] was

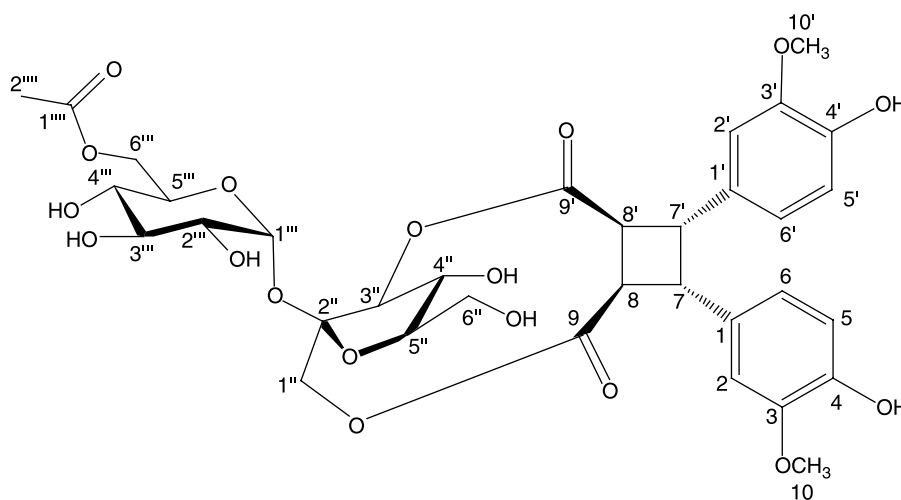


Figure 1. Chemical structure of compound **1**.

Table 1. ¹H NMR and ¹³C NMR spectral data, and HMBC correlations of compound **1** in pyridine-*d*₅.

Aglycone moiety				Sucrose moiety			
No.	δ_C (multiplicity)	δ_H (coupling pattern, J in Hz)	HMBC (<i>H</i> to <i>C</i>)	No.	δ_C (multiplicity)	δ_H (coupling pattern, <i>J</i> in Hz)	HMBC (<i>H</i> to <i>C</i>)
1	130.5 (C)	–		1 ^{'''}	94.2 (CH)	6.18 (d, 3.2)	C-2 ^{'''} , C-3 ^{'''} , C-2 ^{'''}
2	112.7 (CH)	6.76 (br s)	C-1, C-3	2 ^{'''}	73.5 (CH)	4.22 (dd, 3.2, 8.8)	C-3 ^{'''} , C-4 ^{'''}
3	148.3 (C)	–		3 ^{'''}	75.4 (CH)	4.78 (dd, 8.8, 8.4)	C-2 ^{'''} , C-4 ^{'''}
4	146.7 (C)	–		4 ^{'''}	71.9 (CH)	4.28 (dd, 8.4, 8.8)	C-3 ^{'''} , C-5 ^{'''}
5	116.1 (CH)	7.03 (d, 8.0)	C-4, C-6	5 ^{'''}	71.3 (CH)	5.24 (ddd, 8.8, 3.2, 4.4)	C-3 ^{'''} , C-4 ^{'''} , C-6 ^{'''}
6	121.5 (CH)	6.94 (br d, 8.0)	C-5, C-1	6 ^{'''}	64.8 (CH ₂)	4.93 (dd, 4.4, 12.0)	C-5 ^{'''} , C-1 ^{'''}
7	45.8 (CH)	4.57 (dd, 5.2, 9.6)	C-1, C-8, C-7'	1''	65.0 (CH ₂)	5.26 (dd, 3.2, 12.0)	C-4'', C-9
8	43.9 (CH)	4.49 (dd, 5.2, 11.2)	C-7, C-8', C-9	2''	109.6 (C)	–	
9	174.5 (C)	–		3''	79.7 (CH)	5.68 (s)	C-2'', C-4'', C-5'', C-9'
10	55.8 (CH ₃)	3.73 (3H, s)	C-3	4''	88.8 (CH)	4.66 (br s)	C-2'', C-3'', C-5'', C-1''
1'	130.3 (C)	–		5''	75.0 (CH)	5.13 (dd, 3.8, 5.6)	C-2'', C-4'', C-6''
2'	112.7 (CH)	6.88 (br s)	C-1', C-3'	6''	63.3 (CH ₂)	4.47 (dd, 3.8, 12.4)	C-4'', C-5''
3'	148.1 (C)	–		1 ^{''''}	171.1 (C)	–	
4'	146.6 (C)	–		2 ^{''''}	20.7 (CH ₃)	1.88 (3H, s)	C-1 ^{''''}
5'	115.9 (CH)	6.97 (d, 8.0)	C-4', C-6'				
6'	121.1 (CH)	6.71 (br d, 8.0)	C-5', C-1'				
7'	44.8 (CH)	4.80 (dd, 5.2, 9.6)	C-7, C-8', C-1'				
8'	44.5 (CH)	4.37 (dd, 5.2, 11.2)	C-7', C-8, C-9'				
9'	171.9 (C)	–					
10'	55.7 (CH ₃)	3.61 (3H, s)	C-3'				

applied to a silica gel (Merck) column (10 × 60 cm) chromatography and eluted with CHCl₃/MeOH/H₂O (10:3:1) resulting in six fractions monitored by TLC. Fraction 3 (250 mg, V_e/V_t 0.45–0.60) was applied to an ODS (Merck) column (2.5 × 30 cm) chromatography eluted with MeOH/H₂O (3:1), yielding compound **1** [33 mg, V_e/V_t 0.53–0.60, TLC (RP-18 F₂₅₄) R_f 0.55 in MeOH/H₂O = 5:1].

3.3.1 *Impecyloside (1)*

Yellow amorphous powder (MeOH); $[\alpha]_D^{25} + 12$ (c 0.01, MeOH); IR ν_{\max} (CaF₂ window in MeOH) cm⁻¹: 3400, 2937, 1730, 1600, 1453, 1430, 1365; UV λ_{\max} (MeOH): 212 nm; FAB-MS (negative) *m/z*: 735 [M - H]⁻, 693, 559, 453, 421, 405, 271, 183, 175, 133; HRFAB-MS (negative) *m/z*: 735.2139 [M - H]⁻ (calcd. for C₃₄H₃₉O₁₈, 735.2136); NMR spectral data: see Table 1.

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References

- ¹ J.H. Park, *Medicinal Plants of Korea* (Shinil Publishing Co., Seoul, 2004), p. 101.
- ² J.S. Yoon, M.K. Lee, S.H. Sung, and Y.C. Kim, *J. Nat. Prod.* **69**, 290 (2006).
- ³ K. Nishimoto, M. Ito, and S. Natori, *Tetrahedron* **24**, 725 (1968).
- ⁴ K. Matsunaga, M. Ikeda, M. Shibuya, and Y. Ohizuni, *J. Nat. Prod.* **57**, 1290 (1994).
- ⁵ K. Matsunaga, M. Shibuya, and Y. Ohizuni, *J. Nat. Prod.* **57**, 1183 (1994).
- ⁶ M. Wang, S. Wang, Q. Sun, and L. Wu, *J. Chin. Pharm. Sci.* **5**, 53 (1996).
- ⁷ G. Descotes, G. Muller, and J. Mentech, *Carbohydr. Res.* **134**, 313–323 (1984).
- ⁸ Y.G. Goyda, A.F. Abdel-Baky, K.M. Mohamed, F.M. Darwish, R. Kasai, and K. Yamasaki, *Nat. Prod. Res.* **20**, 935 (2006).
- ⁹ C. Ceorgopoulou, N. Aligiannis, N. Fokialakis, S. Mitaku, and J. Asian, *Nat. Prod. Res.* **7**, 799 (2005).
- ¹⁰ L.H. Dimberg, R.E. Andersson, S. Gohil, S. Bryngelsson, and L.N. Lundgren, *Phytochemistry* **56**, 843 (2001).
- ¹¹ Y.M. Chi, M. Nakamura, X.Y. Zhao, T. Yoshizawa, W.M. Yan, F. Hashimoto, J. Kinzo, T. Nohara, and S. Sakurada, *Biol. Pharm. Bull.* **29**, 580 (2006).